

FORM P1

REPUBLIC OF SOUTH AFRICA  
PATENTS ACT, 1978  
APPLICATION FOR A PATENT AND ACKNOWLEDGEMENT OF RECEIPT  
(Section 30(1) - Regulation 39)

REPUBLIC OF SOUTH AFRICA  
REVENUE

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The grant of a Patent is hereby requested by the undermentioned applicant(s) on the present application in duplicate

MT 1669

21	01	Official Application No	22	Lodging Date	47	J & K Reference
9610519			13 December 1996		AP 32488 ZA	
71	Full name(s) of applicant(s)					
BAYER AKTIENGESELLSCHAFT, a legal body organised and existing under the laws of Federal Republic of Germany						
	Address(es) of applicant(s)					
Leverkusen, D-51368, Federal Republic of Germany						
54	Title of Invention					
PROCESS FOR THE PREPARATION OF SYNTHETIC PYRETHROIDS BY AZEOTROPIC ESTERIFICATION						

<input checked="" type="checkbox"/>	The applicant claims priority as set out in the accompanying form P2. The earliest priority claimed is DE 195 46 920.8 15 December 1995	24	01	
<input type="checkbox"/>	This application is for a Patent of Addition to Patent Application No.	21	01	
<input type="checkbox"/>	This application is a fresh application in terms of S 37 and based on Application No.			
<input type="checkbox"/>	This application is accompanied by:			
<input checked="" type="checkbox"/>	1a A single copy of a provisional specification of pages			
<input type="checkbox"/>	1b Two copies of a complete specification of 8 pages			
<input type="checkbox"/>	2a Informal drawings of sheets			
<input type="checkbox"/>	2b Formal drawings of sheets			
<input checked="" type="checkbox"/>	3. Publication particulars and abstract (form P8 in duplicate)			
<input type="checkbox"/>	4. A copy of Figure of the drawings for the abstract			
<input checked="" type="checkbox"/>	5. Assignment of invention (from the Inventors) or other evidence of title			
<input checked="" type="checkbox"/>	6. Certified priority documents (1 documents)			
<input checked="" type="checkbox"/>	7. Translation of priority documents (1 documents)			
<input type="checkbox"/>	8. Assignment of priority rights			
<input type="checkbox"/>	9. A copy of form P2 and the specification of S.A. Patent Application	21	01	
<input checked="" type="checkbox"/>	10. A declaration and power of attorney on form P3			
<input type="checkbox"/>	11. Request for ante-dating on form P4			
<input type="checkbox"/>	12. Request for classification on form P9			
<input type="checkbox"/>	13a Request for delay of acceptance on form P4			
<input type="checkbox"/>	13b			

74 Address for Service: JOHN & KERNICK, Midrand.

Date 13 December 1996

For the Applicant

The duplicate will be returned to the applicant for service as proof of lodging but is not valid unless endorsed with official stamp.

Official date stamp  
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13 DEC 1996

Registrar of patents  
REGISTRAR VAN PATENTE, MODELLE,  
HANDELSMERKE EN OUTEKERSKE

FORM P7

REPUBLIC OF SOUTH AFRICA  
PATENTS ACT, 1978

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**COMPLETE SPECIFICATION**

(Section 30(1) - Regulation 28)

21	01	Official Application No <b>9610519</b>	22	Lodging Date 13th December 1996	47	J & K Reference AP 32488 ZA
51	International Classification C07C					
71	Full name(s) of applicant(s) <b>BAYER AKTIENGESELLSCHAFT, a legal body organised and existing under the laws of Federal Republic of Germany</b>					
72	Full name(s) of inventor(s) <b>Mattias DECKER, Michael ESSER, Martin LITTMANN, Hans-Peter SEHNEM</b>					
54	Title of Invention <b>PROCESS FOR THE PREPARATION OF SYNTHETIC PYRETHROIDS BY AZEOTROPIC ESTERIFICATION</b>					

(I)

R<sup>1</sup> represents methyl, difluoromethyl, trifluoromethyl or chlorine and

Compounds of this type have achieved importance, in particular in controlling insects.

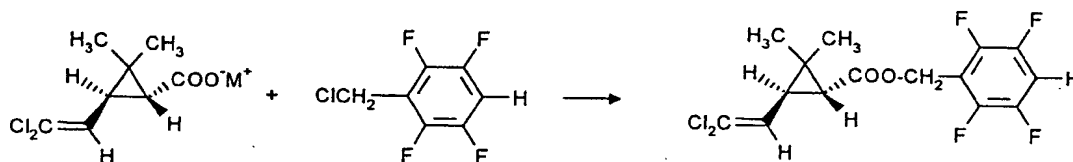
CC1(C)C(Cl)C(Cl)=C[C@H]1C(=O)Cl.Oc1cc(F)c(F)c(F)c1>>CC1(C)C(Cl)C(Cl)=C[C@H]1C(=O)OCc2cc(F)c(F)c(F)c2

20 In the process, the D-menthyl ester of (+)-trans-permethric acid is saponified in methanol/sodium hydroxide solution. After the methanol is distilled off, water is added and the D-menthol is extracted with toluene. After acidification of the aqueous phase, the (+)-trans-permethric acid is extracted with toluene.

The toluene solution is distilled until it is free of water and then thionylchloride is added. Toluene and excess thionyl chloride are then distilled off; (+)-trans-permethrly chloride remains in the bottom phase.

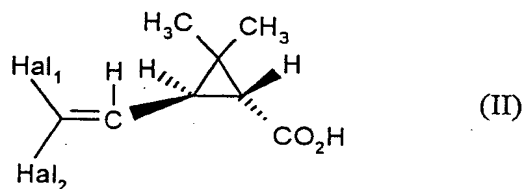
5 The acid chloride is pumped to the tetrafluorobenzyl alcohol introduced; reacting to form transfluthrin. The product is firstly subjected to a steam distillation, then diluted with toluene and the toluene solution is stirred with potassium hydroxide solution to hydrolyze the permethric anhydride. After phase separation, toluene is distilled off and the product can be drawn off.

10 DE 37 05 224 further discloses that the active compound 2,3,5,6-tetrafluorobenzyl-(+)-1R-trans-2,2,-dimethyl-3-(2,2-dichlorovinyl)-cyclopropane-carboxylate (trans-fluthrin) is obtained by reaction of the salts of (+)-trans-permethric acid with 2,3,5,6-tetrafluorobenzyl chloride according to the following scheme:



15 For example, the sodium salt of permethric acid is used, and this is reacted with 2,3,5,6-tetrafluorobenzyl chloride, obtainable from the corresponding benzyl alcohol with, for example, SOCl<sub>2</sub>.

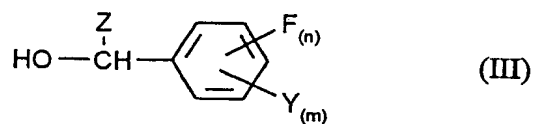
Finally, EP 378 026 generally discloses a process by which compounds of the formula (II)



20 in which

Hal<sub>1</sub> and Hal<sub>2</sub> can be identical or different and represent fluorine, chlorine, bromine or iodine,

are esterified with benzyl alcohols of the formula



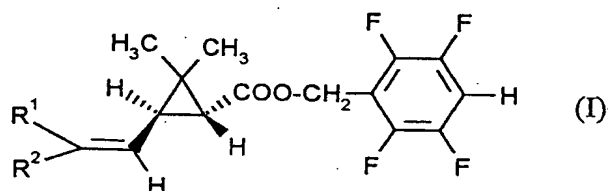
in which

n represents 1 to 5 and

Y can represent, inter alia, hydrogen.

Disadvantages of all of the processes are the complex preparation of the components to be used, which usually have to be prepared in preliminary stages, and the accordingly low total yield of synthetic pyrethroid. A further disadvantage of the process described in the prior art is the occurrence of toxic byproducts, in particular the anhydride of the carboxylic acid used in the particular case.

It has now been found that synthetic pyrethroids of the general formula (I)

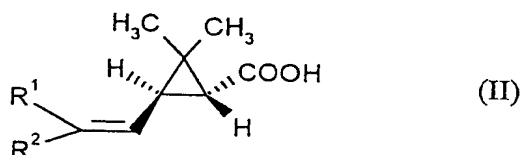


in which

R<sup>1</sup> represents methyl, difluoromethyl, trifluoromethyl or chlorine and

R<sup>2</sup> represents methyl, difluoromethyl, trifluoromethyl or chlorine,

can be prepared by azeotropic esterification of corresponding carboxylic acids of the general formula (II)



in which

$R^1$  and  $R^2$  have the meanings given above, with 2,3,5,6-tetrafluorobenzyl alcohol.

5 The advantages of this process are based in the simple procedure. The carboxylic acids of the formula (II) need not be converted to the acid chloride, or 2,3,5,6-tetrafluorobenzyl alcohol need not be converted to the corresponding chloride. In addition, during the reaction, anhydrides of the corresponding carboxylic acid are not formed, which generally have unfavorable properties and therefore need to be disposed of with high expenditure.

10 The process of the invention is suitable in particular for the synthesis of trans-fluthrin (= 2,3,5,6-tetrafluorobenzyl-(+)-IR-trans-2,2dimethyl-3-(2,2-dichlorovinyl)-cyclopropane-carboxylate).

15 In the process, the toluene solution of (+)-trans-permethric acid is introduced, 5 to 7 mol% of sulfuric acid is added and the mixture is heated to reflux. With azeotropic removal of water, 80 mol% of tetrafluorobenzyl alcohol are added and the reaction mixture is kept at reflux for a further 4 h.

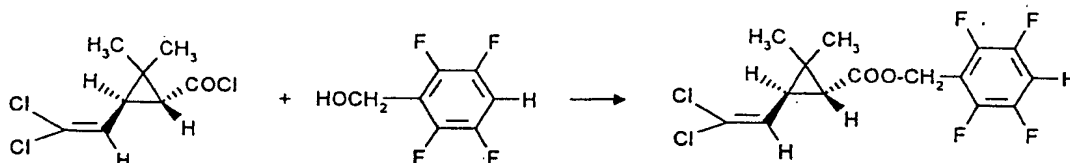
The reaction solution is cooled, the excess acid is extracted as sodium salt with sodium hydroxide solution and reused after the saponification.

Finally, the toluene is distilled off and the active compound remaining in the bottom phase is drawn off.

20 By means of the present azeotropic esterification process of the invention, on the one hand the yield is increased, and furthermore, shorter cycle times are achieved and thus, overall, higher monthly yields.

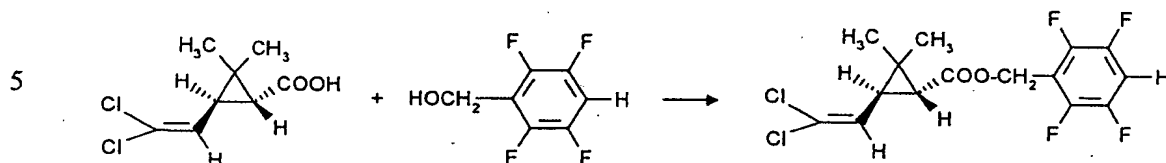
A comparison of the processes makes this clear:

Previous process (A) according to DE 37 05 224



Yield 83%, cycle time 38 hours with batchwise procedure, capacity 18.5 tonnes/month.

Process of the invention (azeotropic distillation); example transfluthrin



Yield 93%, cycle time 18 hours with batchwise procedure, capacity 22 tonnes/month.

10 The process of the invention is carried out in the presence of a solvent suitable for azeotropic esterification processes. Examples of these solvents which may be mentioned are toluene and benzene.

The reaction temperatures depend in this process on the boiling point of the solvent to be used.

The process of the invention is preferably carried out at atmospheric pressure.

15 The process of the invention is preferably carried out with a catalyst. The catalysts used are preferably p-toluenesulfonic acid, acidic ion exchangers or sulfuric acid, but in particular sulfuric acid.

20 In the process of the invention, equimolar amounts of carboxylic acid and 2,3,5,6-tetrafluorobenzyl alcohol are preferably used. However, it has been found that a one to 1.5-fold excess of carboxylic acid has an absolutely beneficial effect on the course of the process (molar ratio: 1.5 mol of carboxylic acid: 1 mol of 2,3,5,6-tetrafluorobenzyl alcohol).

**Experimental part**

376.2 g (1.8 mol) of (+)-trans-permethric acid are introduced in 860 g of toluene and 8 g of sulfuric acid are added. The batch is heated to reflux and in the course of approximately 1 h, 270 g (1.5 mol) of 2,3,5,6-tetrafluorobenzyl alcohol in 116.0 g of toluene are added; water is simultaneously removed azeotropically. The batch is kept at reflux for a further 4 h, the reaction water being removed azeotropically.

The reaction mixture is cooled to room temperature and extracted with 300 ml of 2 molar sodium hydroxide solution to remove the excess acid.

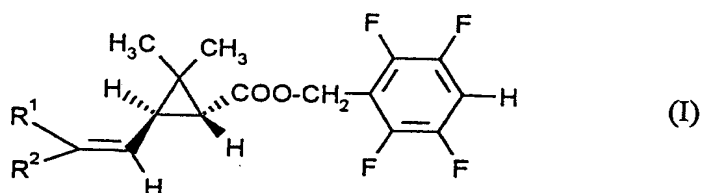
- 10 The organic phase is washed with 500 ml of water and freed from solvent. (The residue is mixed with 250 ml of water and steam-volatile minor components are removed by distilling off water).

Yield: 539.6 g  $\cong$  97.0% transfluthrin



# Patent Claims

1. Process for the preparation of synthetic pyrethroids of the general formula (I)

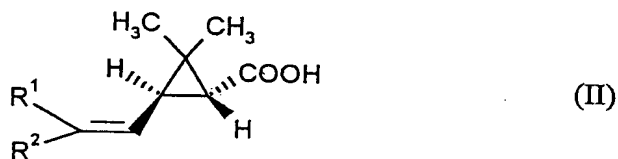


5 in which

R<sup>1</sup> represents methyl, difluoromethyl, trifluoromethyl or chlorine and

R<sup>2</sup> represents methyl, difluoromethyl, trifluoromethyl or chlorine,

characterized in that compounds of the formula (II)



10 in which

R<sup>1</sup> and R<sup>2</sup> have the meanings given above,

are azeotropically esterified with 2,3,5,6-tetrafluorobenzyl alcohol.

2. Process for the preparation of 2,3,5,6-tetrafluorobenzyl-(+)-1R-trans-2,2-dimethyl-3-(2,2-dichlorovinyl)cyclopropane-carboxylate by azeotropic esterification of (+)-trans-permethric acid with 2,3,5,6-tetrafluorobenzyl alcohol.
3. Process for the preparation of synthetic pyrethroids according to claim 1, characterized in that solvents suitable for azeotropic esterifications are used.

4. Process for the preparation of synthetic pyrethroids according to claim 1, characterized in that the solvent used is toluene.
5. Process for the preparation of synthetic pyrethroids according to claim 1, characterized in that it is carried out under atmospheric pressure.
- 5 6. Process for the preparation of synthetic pyrethroids according to claim 1, characterized in that 1 to 1.5 mol of carboxylic acid are azeotropically esterified with 1 mol of 2,3,5,6-tetrafluorobenzyl alcohol.
7. Process for the preparation of synthetic pyrethroids according to claims 1  
10 to 6, characterized in that the process product is free of anhydrides of the carboxylic acids respectively used.
8. Process for the preparation of 2,3,5,6-tetrafluorobenzyl-(+)-1R-trans-2,2-  
dimethyl-3-(2,2-dichlorovinyl)-cyclopropane-carboxylate, characterized in  
that a mixture of (+)-trans-permethric acid, toluene and sulfuric acid is  
15 heated to reflux, a mixture of 2,3,5,6-tetrafluorobenzyl alcohol in toluene is added, the whole batch is azeotropically esterified finally the reaction mixture is cooled to room temperature and excess acid and solvent are removed.
9. Process for the preparation of synthetic pyrethroids of the general formula (I) substantially as herein described and as exemplified with reference to the Experimental Part.

DATED THIS 13TH DAY OF DECEMBER 1996

  
JOHN & KERNICK  
FOR THE APPLICANT